Analysis Activity

Did the radius increase or

decrease?

decrease

Directions:

ION FORMATION

1. Log into Collisions and navigate to the lons Game.



Name: _

2. Play the Tutorial lev	els, if you have	n't done so alre	eady.						
3. Exit the levels and	enter the lons S	Sandbox. 🚹							
4. Ionize each atom li	sted in the table	e below and red	cord the reques	ted informatior	١.				
_			_			_			
	Li	Ве	В	N	0	F	Na	Mg	Al
STEP 1: Drag the ato	m into the worl	kspace and rec	ord the followin	g information.					
# of protons	3								
# of electrons	3								
STEP 2: Form an ion I complete area. Once		-	-	-					
# of LOST or GAINED electrons	ı lost								
# of electrons in ions	2								
lon charge	+1								
Amount of energy USED or	6 units used								

ION FORMATION



	Р	S	Cl	К	Ca	As	Se	Br	
STEP 1: Drag the atom into the workspace and record the following information.									
# of protons									
# of electrons									
STEP 2: Form an ion by adding or removing electrons. When you think you have created an ion, hit the CHECK button. If correct, the ion will move into the complete area. Once you have successfully created the ion, drag it from the complete area back into the workspace to record the following information.									
# of LOST or GAINED electrons									
# of electrons in ions									
Ion charge									
Amount of energy USED or RELEASED?									
Did the radius increase or decrease?									

Analysis Questions:

- 1. Look at your recorded energy used and released. Write a summary of your observations below.
- $2. \ Look\ at\ your\ recorded\ radius\ increase\ or\ decrease.\ Write\ a\ summary\ of\ your\ observations\ below.$
- 3. On a separate sheet of paper, create a graph of atomic number vs. energy used for all positive ions above.